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Ideaalikaasun tiheys.

Lämpölaajenemisen vuoksi samankin aineen tiheys riippuu lämpötilasta.

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Kiinteillä aineilla ja nesteillä lämpölaajeneminen on kuitenkin melko vähäistä, mutta kaasuilla paljon suurempaa, jota paitsi niiden tiheys riippuu myös paineesta. Ideaalikaasun tiheys (ρ) riippuu sen moolimassasta ...

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foundation in understanding both human-caused climate change and natural variability, and has made me a better reporter.”

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Termi ja nimeäminen. Ensimmäisen

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kerran isotooppi-termiä käytti vuonna 1913 Margaret Todd, joka ehdotti nimeä Frederick Soddyille. Soddy huomasi, että eräillä alkuaineilla on useita isotooppeja, jotka kuitenkin kuuluvat jaksollisessa järjestelmässä samaan paikkaan. Tämän vuoksi Todd ehdotti nimeksi kreikankielistä sanaa, joka tarkoittaa samassa paikassa.

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